

***The Effect Of Temperature On Light Intensity Of
Different Coloured Glow Sticks***

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Submitted by

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ARRAHMAAN
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The Effect Of Temperature On Light Intensity Of Different Coloured Glow Sticks

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The Effect Of Temperature On Light Intensity Of Different Coloured Glow Sticks

ABSTRACT

Light energy is a form of electromagnetic radiation. Light consists of photons, which are produced when an object's atoms heat up. Light travels in waves and is the only form of energy visible to the human eye. It is possible to produce light without heat or electricity. **Chemiluminescence** is the emission of light caused by a chemical reaction. This type of luminescence can be observed by bending and shaking a light stick.

I wanted to visualize how temperature impacts the light intensity of glowing sticks. My project aims at studying the relationship between the colour of the glow-stick with the light intensity at three different temperatures. For this purpose, I had used the commercial glow-sticks which I bought from Amazon. I had chosen five different colours as White, Pink, Lime-green, Yellow and Orange for my study. For measuring the intensity of light produced, I constructed the light measuring device by making use of Light Detecting Resistor (LDR) also known as photoresistor. I completely wrapped three glass jars using aluminium foil leaving the open end, in order to ensure no light is escaped. I inserted the photo-resistor through the hole made in the three lids and well-insulated them. The resistance exhibited by the photoresistor is the measure of the light intensity produced from the glow-sticks. I connected the exposed leads of photo-resistor with the multimeter. Before proceeding, the photoresistor was tested for its baseline in open lid and closed lid condition. Then, I started twisting the glow-sticks and putting inside the empty glass jar setups to measure the light intensity at room temperature which is followed by ice-cold water and hot water baths for all the five colours. At a time, I have done three trials for all the cases. The multimeter reading which shows the resistance in kilo-ohms ($k\Omega$) was taken at every minute for 15 minutes. The average readings were tabulated and plotted into graph. I observed the greater increase in the photo-resistance in ice cold water from that of hot and room temperatures, which implies low light intensity. I formulated that ***increasing temperature actually increases light intensity*** unless and until the heat is maintained. Comparing all colours, ***Yellow has more light intensity in all the temperatures.***



INTRODUCTION

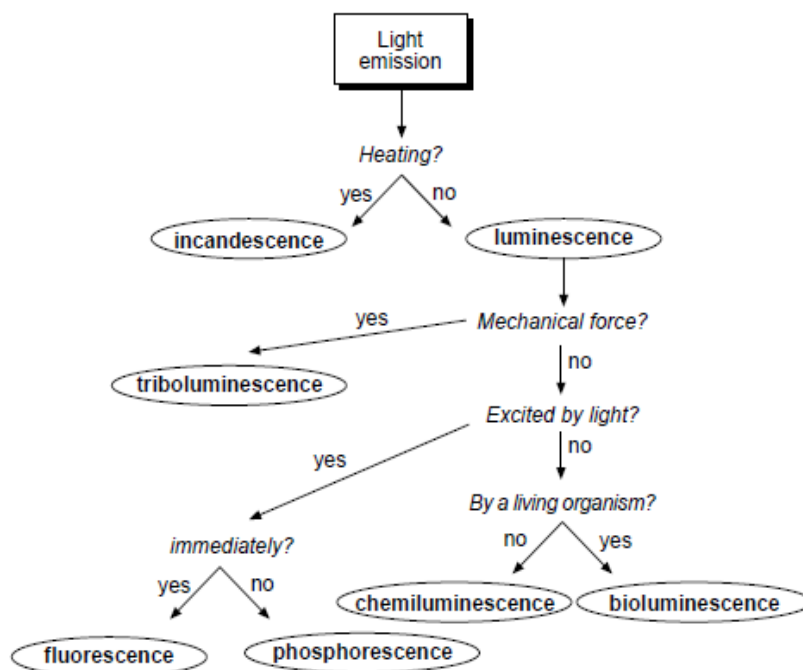
Light energy is used to help us see – either naturally using the Sun or fire, or with manmade objects like candles or light bulbs. Light energy is also used by plants, which capture the light energy from the Sun and use it to produce their food. Light is made up of photons, which are like tiny packets of energy. When an object's atoms heat up, photons are produced from the movement of atoms. The hotter the object, the more photons are produced. The word “photon” comes from Latin word photo – which means light.



Figure 1: Glow Worms are capable for producing their own light - Bioluminescence

Cold light

Cold light is also known as luminescence. It is light that is emitted when things are cold. Light emitted when things get hot is called incandescence – for example, the light from a Bunsen burner, a standard light bulb or an electric cooker when hot. There are many types of cold light, given different names depending on the conditions under which the light is emitted.



How glow sticks glow?

All liquid glow products depend on a chemical process known as Chemiluminescence to produce their light. Chemiluminescence is a chemical reaction that causes a release of energy in the form of light. To produce this light the electrons in the chemicals become excited and rise to a higher energy level. To utilise this process glow-sticks contain two liquids; hydrogen peroxide and tert-butyl alcohol. When mixed together it is these liquids that create the glow. Fluorescent dyes are also used in the alcohol to alter the colour of the light emitted.

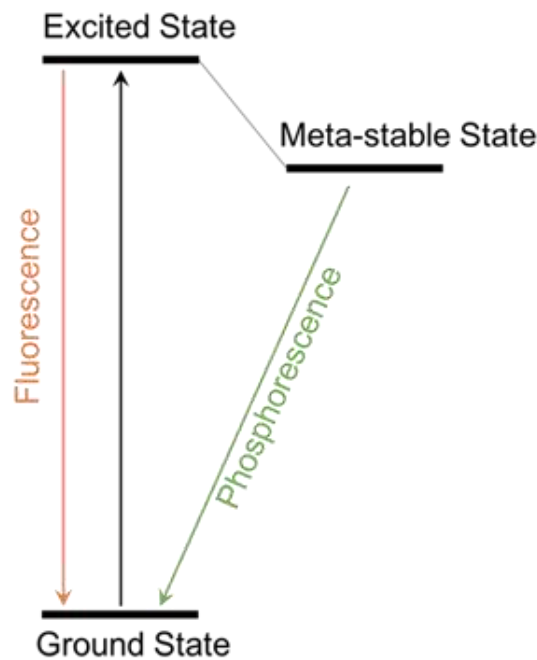





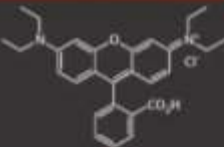






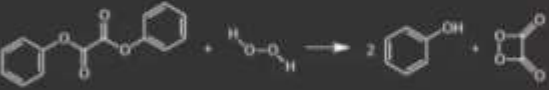
Figure 1: Energy scheme that shows phosphorescence and fluorescence.


When photons interact with a material, electrons can be excited from the ground state to an excited state. When they return to the ground state, they release their energy in the form of light (fluorescence). Electrons can also be trapped in a meta-stable state and when these electrons relax to the ground state they release light by phosphorescence.

When the stick bends, the glass vial breaks allowing the two liquids to mix together. One of the liquids is contained in a very fine glass tube that floats within the mixture inside the plastic glow stick product. This is why you must bend a glow stick to make it start glowing. When the stick bends, the glass vial breaks allowing the two liquids to mix together. The chemical reaction begins immediately resulting in a bright, fantastic glow.

THE CHEMISTRY OF GLOW STICKS

				
RED	ORANGE	YELLOW	GREEN	BLUE
				
RHODAMINE B	1,12-DIPHENYLETHYNYLANTHRACENE	RUBRENE	6,20-BIS(PHENYLETHYNYL)ANTHRACENE	9,10-DIPHENYLANTHRACENE






HOW DO GLOW STICKS PRODUCE LIGHT?

When glow sticks are bent, the inner glass tube is broken, releasing hydrogen peroxide solution. This then reacts with a diphenyl oxalate, producing 1,2-dioxetanedione; this product is unstable, & decomposes to carbon dioxide, releasing energy. The energy is absorbed by electrons in dye molecules, which subsequently fall back to their ground state, losing excess energy in the form of light.

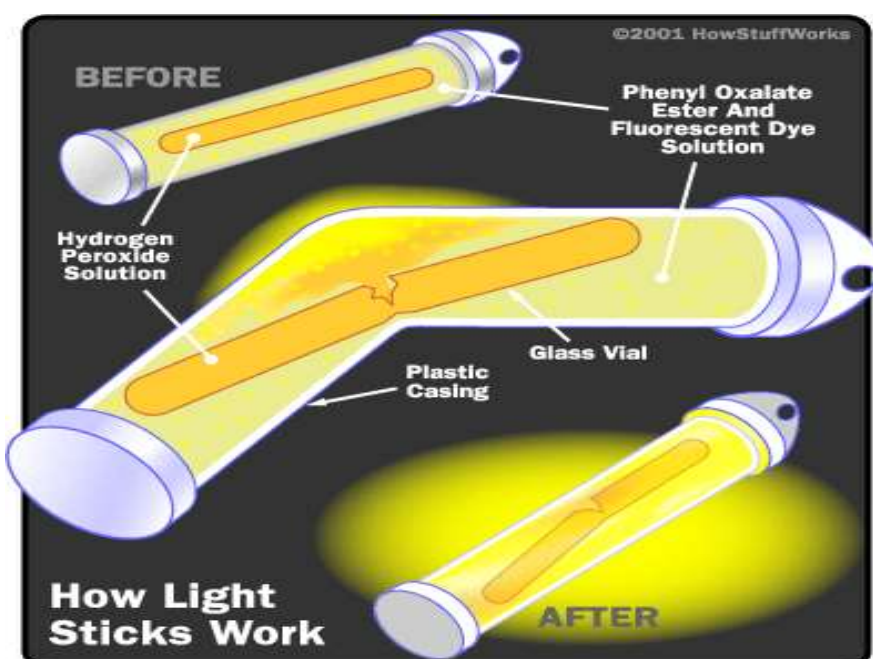
DIPHENYL
OXALATE
ESTER



HYDROGEN
PEROXIDE

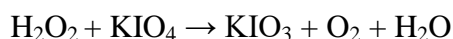
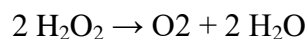
How long does a glow stick glow for?

The duration of glow stick's glow will vary depending upon the type of liquids used, the exact composition and also the quality of the liquids inside. Temperature also affects the intensity of the glow - the warmer it is the brighter the glow but this shortens the total glow duration. The colder it is the glow stick will glow less but will glow for a longer duration.



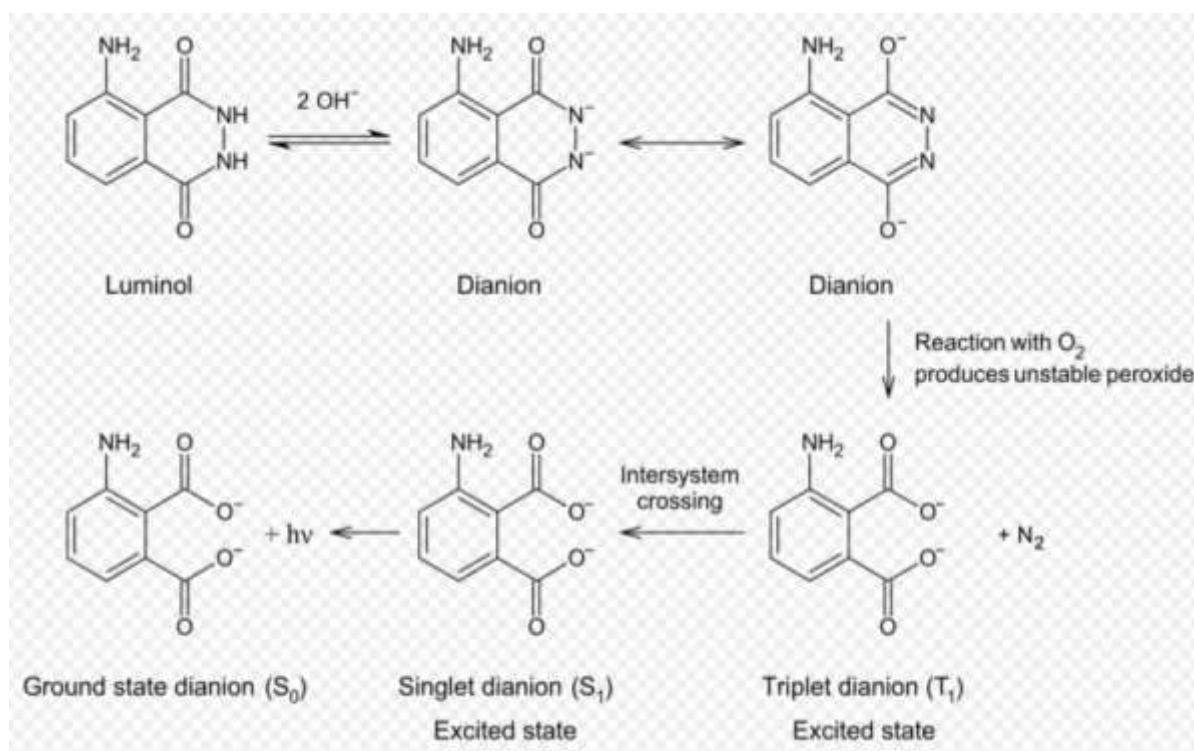
Chemiluminescence of luminol

To exhibit its luminescence, the luminol must be activated with an oxidant. Usually, a solution containing hydrogen peroxide (H_2O_2) and hydroxide ions in water is the activator. In the presence of a catalyst such as an iron or periodate compound, the hydrogen peroxide decomposes to form oxygen and water:



Laboratory settings often use potassium ferri-cyanide or potassium periodate for the catalyst. In the forensic detection of blood, the catalyst is the iron present in haemoglobin. Enzymes in a variety of biological systems may also catalyse the decomposition of hydrogen peroxide.

Luminol reacts with the hydroxide ion, forming a di-anion. The oxygen produced from the hydrogen peroxide then reacts with the luminol dianion. The product of this reaction, organic peroxide, is very unstable and immediately decomposes with the loss of nitrogen to produce 5-aminophthalic acid with electrons in an excited state. As the excited state relaxes to the ground state, the excess energy is liberated as a photon visible as blue light.



Similarities and differences between chemi-luminescence and phosphorescence:

Fluorescence and phosphorescence are two mechanisms that emit light or examples of photoluminescence. However, the two terms don't mean the same thing and don't occur the same way. In both fluorescence and phosphorescence, molecules absorb light and emit photons with less energy (longer wavelength), but fluorescence occurs much more quickly than phosphorescence and does not change the spin direction of the electrons.

Fluorescence Versus Phosphorescence

1. Both fluorescence and phosphorescence are forms of photoluminescence. In a sense, both phenomena cause things to glow in the dark. In both cases, electrons absorb energy and release light when they return to a more stable state.
2. Fluorescence occurs much more quickly than phosphorescence. When the source of excitation is removed, the glow almost immediately ceases (fraction of a second). The direction of the electron spin does not change.
3. Phosphorescence lasts much longer than fluorescence (minutes to several hours). The direction of the electron spin may change when the electron moves to a lower energy state.

Examples of Phosphorescence

Phosphorescent materials are used in gun sight, glow-in-the dark stars, and paint used to make star murals.

The element phosphorus glows in the dark, but not from phosphorescence.

Luminescence is the emission of light produced by methods other than heat. Luminescence is caused by the movement of electrons into different energetic states. There are many different types of luminescence including bioluminescence, chemiluminescence, phosphorescence, and fluorescence. These various forms of luminescence differ in their method of emitting light. Fluorescent and phosphorescence are only two ways light may be emitted from a material.

Chemiluminescence is the emission of light caused by a chemical reaction. *Chemiluminescence is a chemical reaction that causes a release of energy in the form of light.* This type of luminescence can be observed by bending and shaking a light stick. Inside the light stick is an encapsulated chemical solution, surrounded by a different chemical solution. Bending the light stick causes the encapsulated chemical solution to break open. Shaking the light stick then causes the two solutions inside the stick to mix. When

these chemicals combine, energy is created in the form of light. Organisms that emit light, known as bioluminescent organisms, also produce light through a chemical reaction.

Phosphorescence is the ability of a material to absorb energy from an electromagnetic radiation source, such as a flashlight, and then continue to emit light after the source has been removed. In this activity, the glow in the dark object absorbs or stores the energy from the flashlight and then gradually re-emits the energy as light, even after the flashlight has been turned off.

Fluorescence is similar to phosphorescence, in that it is the ability of a material to emit light by absorbing energy from a source of electromagnetic radiation. However, unlike phosphorescence, fluorescent materials can only emit light during the time that they are exposed to the source of electromagnetic radiation. A fluorescent material will stop “glowing” once the light source is removed.

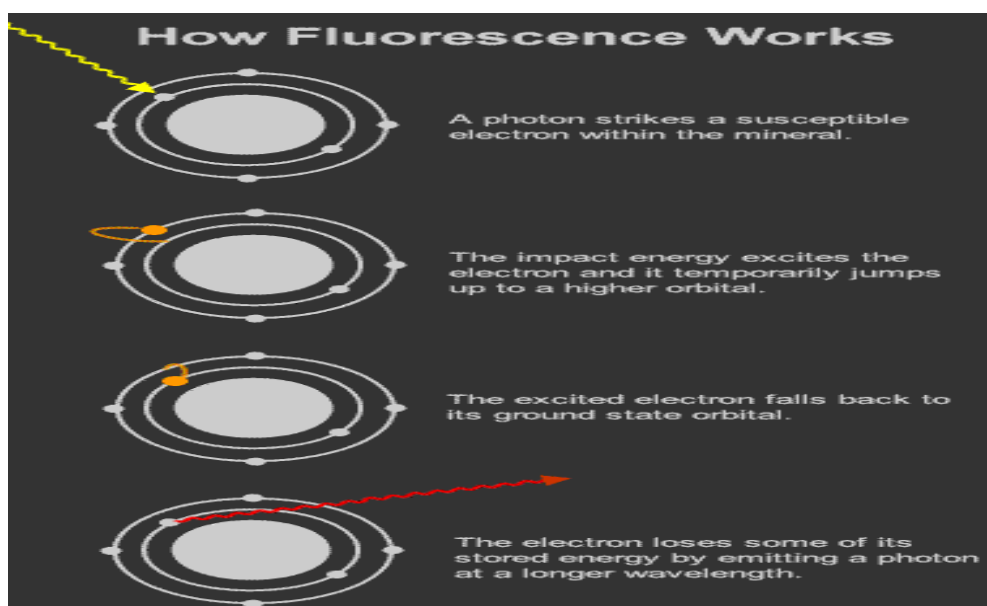
Factors Affecting Fluorescence Efficiency

Structural Rigidity: Collisional deactivation is a major fluorescence quenching mechanism. Therefore, molecules possessing rigid structures are better fluorophores than others which lack rigidity. This explains why fluorene is an excellent fluorophore while biphenyl is a weak one.

Temperature: As temperature is increased, the translational, rotational and vibrational motions of molecules increase. This increases the possibilities of collisions and lead to collisional deactivation and quenching of fluorescence. Therefore, it is always wise to conduct fluorometric measurements at low temperatures

pH: Usually, fluorophores that contain either acidic or basic moieties have fluorescence quantum yields dependent on pH. The pH of these substances should be adjusted so that maximum fluorescence is obtained.

(Future Enhancement)



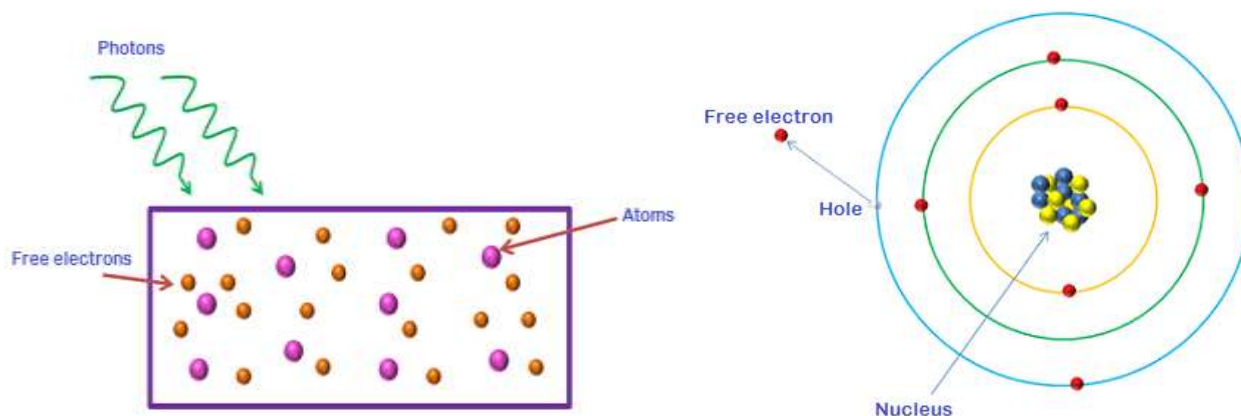
Photoresistor

The name photoresistor is the combination of words: photon (light particles) and resistor. A photoresistor is a type of resistor whose resistance decreases when the intensity of light increases. In other words, the flow of electric current through the photoresistor increases when the intensity of light increases. Photo-resistors are also sometimes referred as LDR (Light Dependent Resistor), semiconductor photoresistor, photoconductor, or photocell. **Photoresistor changes its resistance only when it is exposed to light.**

How photoresistor works?

When the light falls on the photoresistor, some of the valence electrons absorb energy from the light and break the bonding with the atoms. The valence electrons, which break the bonding with the atoms, are called free electrons.

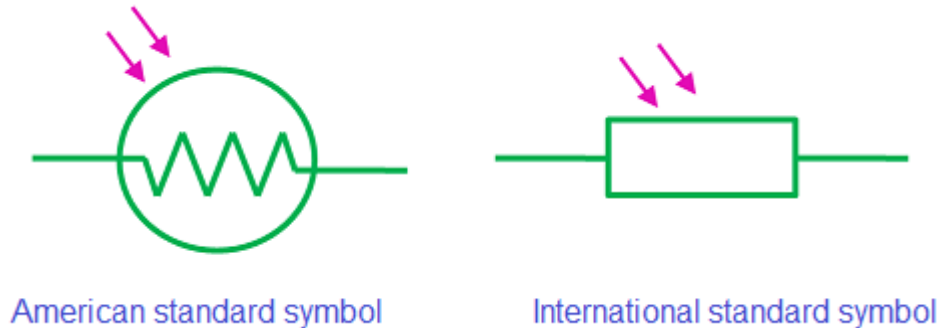
When the light energy applied to the photoresistor is highly increased, a large number of valence electrons gain enough energy from the photons and break the bonding with the parent atoms. The large number of valence electrons, which break the bonding with the parent atoms will jump into the conduction band.



The electrons present in the conduction band do not belong to any atom. Hence, they move freely from one place to another place. The electrons that move freely from one place to another place are called free electrons. When the valence electron left the atom, a vacancy is created at a particular location in an atom from which the electron left. This vacancy is called as hole. Therefore, the free electrons and holes are generated as pairs.

The free electrons that are moving freely from one place to another place carry the electric current. In the similar way, the holes moving in the valence band carry electric current. Likewise, both free electrons and holes will carry electric current. The amount of electric current flowing through the photoresistor depends on the number of charge carriers (free electrons and holes) generated.

When the light energy applied to the photoresistor increases, the number of charge carriers generated in the photoresistor also increases. As a result, the electric current flowing through the photoresistor increases. Increase in electric current means decrease in resistance. Thus, the resistance of the photoresistor decreases when the intensity of applied light increases.



Photoresistors are made of high resistance semiconductor such as silicon or germanium. They are also made of other materials such as cadmium sulfide or cadmium Selenide.

In the absence of light, the photoresistors acts as high resistance materials whereas in the presence of light, the photoresistors acts as low resistance materials.

Types of photoresistors based on material used to construct them

Photoresistors are classified into two types based on the material used to construct them:

- Intrinsic photoresistor
- Extrinsic photoresistor

STATEMENT OF THE PROBLEM

There is no more saying that light can be produced only with electricity. Nature gifts everything that we can't even think of. Light from Sun, firefly and much more we are seeing around gives us a hint of how to make use of them. This is how glow-sticks came into existence. But how extent it may direct way-lost souls in foggy and freezing nights, and which colour lasts long are all a matter of concern now. Hence, it is necessary to know the impact of temperature and colour on how long or brighter will be the glow.

HYPOTHESIS

***Cold temperature increases the Light intensity of
glow sticks.***

DESIGN OF STUDY

INDEPENDENT VARIABLE:

- **Experiment 1:** Temperature
- **Experiment 2:** Colour

DEPENDENT VARIABLES:

- Resistance of Photo Resistor
- Glowing Time

CONTROLLED VARIABLES:

- **Experiment 1:** Length Of The Glow Stick, Amount Of Chemicals Responsible For Chemiluminescence, Colour
- **Experiment 2:** Length Of The Glow Stick, Amount Of Chemicals Responsible For Chemiluminescence, Temperature

MATERIALS:

- Photo-resistor - 3
- Digital multimeter -3
- Alligator clip leads with connecting wires- 3 sets
- Glass jar with lid, big enough to hold glow-in-the-dark-objects -3
- Glow bracelets (Pink, orange, lime-green, yellow and white of length 8”) each 10
- Aluminum foil roll
- Black electrical tape
- Lighting candle and a thin metal wire (for putting hole in the plastic lid)
- Laboratory thermometer
- Liquid measuring cup
- Heat-resistant cup
- Induction stove and a vessel for heating water
- Tap water
- Ice cubes
- Timer or stopwatch
- Lab notebook

PROCEDURE:

- Decide on the surroundings where the setup could be set in.

Building the Light-Measuring Device:

1. In this project, we need to determine the light intensity so that we can compare the longevity and the brightness of the glow with respect to the colour. Build up the Light measuring device with the help of photo-resistor.
2. A photo-resistor or light-dependent resistor, LDR, or photo-conductive cell is a light-controlled variable resistor. The resistance of a photo-resistor decreases with increasing incident light intensity. Photo-resistor is a small, round part, with a squiggly line on the front and two metal leads.
3. Make the light-measuring device by wrapping the glass jar in aluminum foil and taping a photo-resistor in the lid with electrical tape.
 - a. Wrap aluminum foil around the sides of the glass jar so no light gets in and seal all edges securely with black electrical tape.
 - b. Light up the candle and heat one end of a thin metal wire. With the hot wire, considering the resistor head size put two holes nearby in the centre of the plastic lid at once.
 - c. Insert the photo-resistor through the hole in the inside of the lid so that the metal leads could be visible outside. Tape the photo-resistor over the hole in the jar top, so it is facing inside the jar when the lid is closed (the side with the squiggly line should be facing the inside of the jar).
 - d. Leave a small amount of exposed metal at the ends of the leads, so we can attach alligator clips to them later. If the jar lid is made of metal, also cover the metal lid with electrical tape.
 - e. Bend the leads of the photo-resistor sideways and cover them in electrical tape, so they do not bump into each other and create a *short circuit*.

Important: Do not let the leads of the photo-resistor touch the aluminum foil. Make sure to insulate the aluminum foil and the leads with electrical tape; otherwise it will create a short circuit and the resistance of the photo-resistor will always read zero.

Testing the resistance baseline:

1. Determine the resistance baseline for the measuring device and measure the resistance of the photo-resistor for the jar with nothing in it and with the lid on and off.
 - a. Set up the measuring device in a room with as few disturbances in lighting as possible. The photo-resistor is very sensitive, so variations in sunlight coming through a window on a cloudy day, or

even shadows and reflections created as people walk around the room, can all affect the readings if we did not seal the glass jar properly.

- b. Set up the multimeter to measure the resistance of the photo-resistor.
 - i. Plug the black multimeter probe into the port labelled COM.
 - ii. Plug the red multimeter probe into the port labelled $V\Omega mA$.
 - iii. Connect the multimeter probes to the leads of the photo-resistor using alligator clips. Make sure the metal parts of the alligator clips do not touch the aluminum foil, or this will create a short circuit.
 - iv. Set the multimeter dial to measure resistance in the 200 kilo-ohm ($k\Omega$) range.
 - v. Turn the power switch ON.
- c. Leave the lid open and expose the photo-resistor to the surrounding light. Read the resistance across the photo-resistor and record it in the lab notebook.
 - i. Note the units of the resistance. A "k" indicates kilo-ohms ($k\Omega$).
 - ii. If the multimeter screen displays a "1.", that means the resistance is too high for the dial setting. Turn the dial up to the next highest range (for example, from 200 k to 2000 k) and check again.
- d. Close the jar with the photo-resistor lid and read the resistance across the photo-resistor again. In the dark, the resistance should be in the mega-ohm range. Remember that we may need to adjust the dial setting to get a measurement. Record the resistance in your lab notebook. *Note:* The light level in the closed, light-protected (aluminum-foil-wrapped) jar should be the same when the jar is in a lighted room or in a dim room. If it changes, that means there is a light leak. Find the problem area and cover the light leak with electrical tape.
- e. Get ready 2 more glass jars in the same way.

Setting up the glow sticks:

Once the photo-resistor is tested for its baseline, measure the luminescence as a function of resistance.

Measuring Luminescence

Using the light-measuring device, we can measure the light intensity of the glow sticks. To minimize unwanted variations, position the glow sticks in the jar the same way for each trial, keep movement of the setup to a minimum.

STEP I:

1. First, investigate how the light intensity of the glow-in-the-dark object changes over time.

- a. Keep the glowsticks of different colours –White, Pink, Lime-green, Yellow and Orange each 10 in a tray.
- b. First, start with White. Twist three white glow sticks to start the reaction. Make sure to shake it enough so that it glows evenly. Then bend it to form bracelets with the help of connectors.
- c. Place each of three sticks in three aluminum-foil wrapped jar and place the lid on (3 trials). Set the timer to 15 minutes. Note down the starting time also.
- d. Record the time and signal strength produced by the light of the glow stick every minute for 15 minutes.
- e. Follow the same for other four colours each three trials.
- f. After each colour, place the used sticks in a dark room to observe the time period of its glow. Note down the glowing time.

STEP II:

1. Next, investigate how temperature affects the light intensity of the glow-in-the-dark object.
 - a. The first step can be included as the effect of temperature of the light intensity of glowing object **at Room Temperature.**
 - b. Once the 15 minutes are over, open the jar and add **cold water** to decrease the temperature, as follows:
 - i. Prepare a cup of ice cold water by putting a couple of ice cubes in a cup of tap water.
 - ii. Wait for a few minutes so as to attain a temperature range of 5-9°C. then remove the ice cubes.
 - iii. Simultaneously, twist and bend the white sticks readily.
 - iv. Open the jar, and fill the jar with the cold water so that the glowing object is fully submerged.
 - v. Carefully close the jar again.
 - c. Set the timer to 15 minutes and record the resistance of the glow-in-the-dark object in cold water once the 15 minutes are over.
 - d. Follow the same for all the five colours and note down the reading.
 - e. After each colour, keep the used sticks aside in the dark room under observation.

STEP III:

1. Next, open the jar and replace the cold water with hot water to increase the temperature, as follows:
 - i. Prepare a cup of hot water (approximately 70°C-75°C) by heating some water on the stove. Check the temperature using thermometer.

- ii. Carefully open the jar, pour out the cold water and re-fill the jar with the hot water so that the glowing object is fully submerged.
- iii. Carefully close the jar again.
2. Set your timer to 15 minutes and record the resistance of the glow-in-the-dark object in hot water once the 15 minutes are over.
3. Follow the same procedure for 5 different colours (Pink, orange, lime-green, yellow and white) and compare the glowing time of different glow sticks under different temperature.

Analyzing Your Data

1. Calculate the average resistance for each of the time point (every minute) from all three of the trials each of the five colours. Plot the data showing the time, in minutes, on the x-axis, and the average resistance, in kOhms ($k\Omega$), on the y-axis. The resistance curve should start with a low level at first and then gradually rise to a higher level as the light intensity decays. **The resistance of the photo-resistor *increases* with decreasing light intensity.**
2. With the help of the graph, visualize:
 - How the light signal decays with time.
 - How temperature affects the light intensity.
3. Make another data table with the temperature (room temperature, hot, and cold) and the average resistance readings (of three trials) in kOhms ($k\Omega$), of every minute (upto 15 minutes) for each colour. Graph results in a bar graph with the temperature on the x-axis and the resistance on the y-axis.

COLLECTION OF DATA- PHOTOGRAPHS



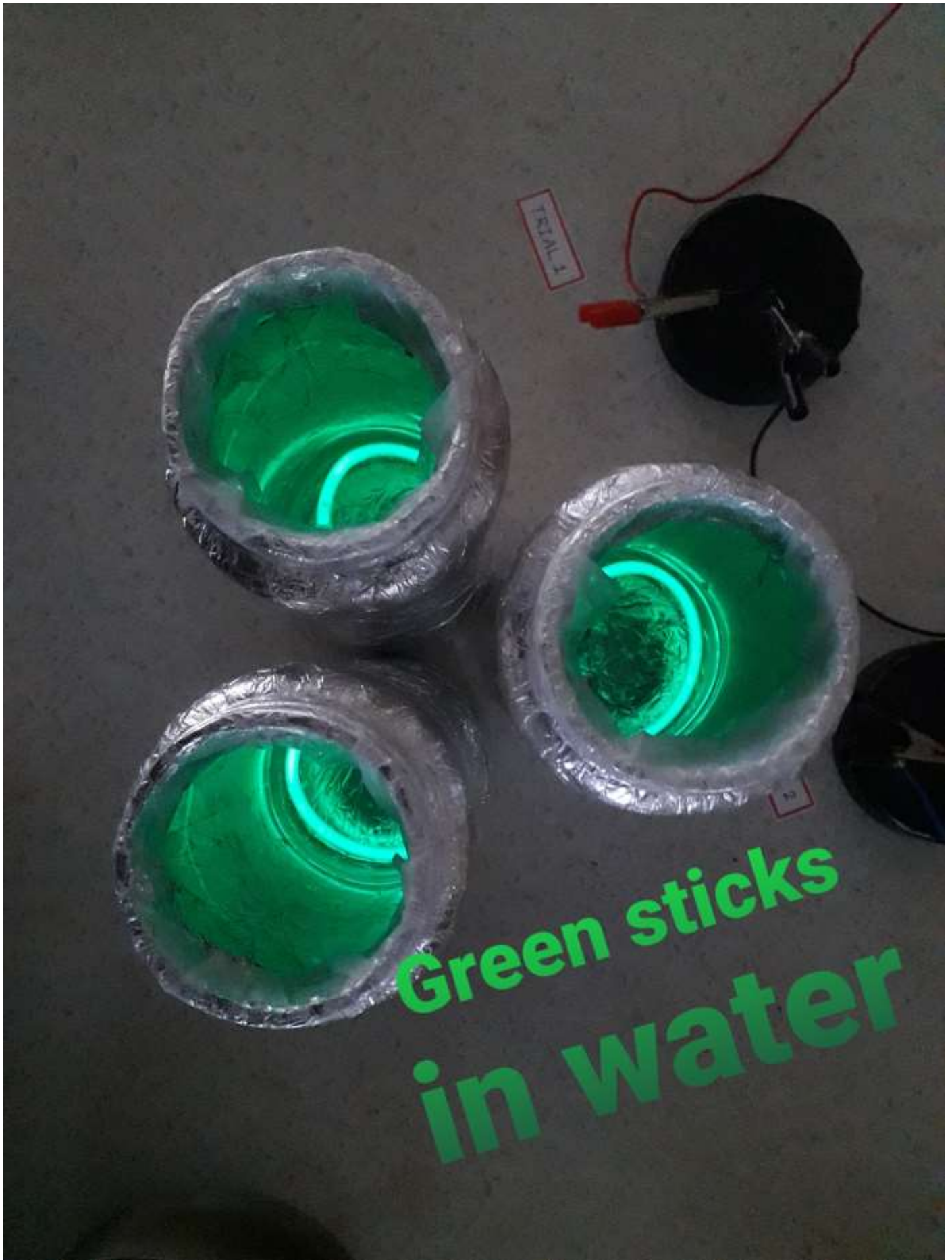




Noting readings







**Cold
water**

Hot water



USED GLOWSTICKS

Used
glowsticks in
dark room





Tabulation

Table1: Measuring the Baseline of the Photo-resistor without glow sticks

	<i>Photo-resistance with open lid (kΩ)</i>	<i>Photo-resistance with closed lid (kΩ)</i>
Trial 1	15	Above 2000
Trial 2	16	
Trial 3	14	

Table 2: Effect of temperature on light intensity of WHITE coloured glow stick

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>											
	<i>At room temperature 31°C</i>				<i>In Ice cold water at 9°C</i>				<i>In Hot water at 75°C</i>			
	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>
0	1379	183	177	579.667	2000	1007	1830	1612.33333	348	111	100	186.333
1	1500	234	242	658.667	2000	1128	1886	1671.33333	525	201	192	306
2	1600	262	296	719.333	2000	1164	1933	1699	616	248	212	358.667
3	1715	284	331	776.667	2000	1189	1975	1721.33333	740	286	240	422
4	1757	305	358	806.667	2000	1217	2000	1739	835	327	272	478
5	1817	318	380	838.333	2000	1245	2000	1748.33333	906	377	303	528.667
6	1861	331	397	863	2000	1269	2000	1756.33333	992	427	333	584
7	1885	342	411	879.333	2000	1281	2000	1760.33333	1084	482	367	644.333
8	1914	352	424	896.667	2000	1307	2000	1769	1171	526	402	699.667
9	1926	361	435	907.333	2000	1326	2000	1775.33333	1296	565	427	762.667
10	2000	368	445	937.667	2000	1342	2000	1780.66667	1360	604	450	804.667
11	2000	377	455	944	2000	1354	2000	1784.66667	1461	660	488	869.667
12	2000	384	463	949	2000	1376	2000	1792	1550	712	519	927
13	2000	390	471	953.667	2000	1387	2000	1795.66667	1681	803	549	1011
14	2000	397	480	959	2000	1397	2000	1799	1773	829	577	1059.67
15	2000	404	488	964	2000	1406	2000	1802	1776	883	606	1088.33

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>		
	<i>At room temperature 31°C</i>	<i>In Ice cold water at 9°C</i>	<i>In Hotwater at 75°C</i>
0	579.667	1612.33333	186.333
1	658.667	1671.33333	306
2	719.333	1699	358.667
3	776.667	1721.33333	422
4	806.667	1739	478
5	838.333	1748.33333	528.667
6	863	1756.33333	584
7	879.333	1760.33333	644.333
8	896.667	1769	699.667
9	907.333	1775.33333	762.667
10	937.667	1780.66667	804.667
11	944	1784.66667	869.667
12	949	1792	927
13	953.667	1795.66667	1011
14	959	1799	1059.67
15	964	1802	1088.33

Table 3 : Effect of temperature on light intensity of PINK coloured glow stick

Time (in minutes)	Photo Resistance (in k Ω)											
	At room temperature 31°C				In Ice cold water at 9°C				In Hot water at 75°C			
	Trial 1	Trial 2	Trial 3	AVG	Trial 1	Trial 2	Trial3	AVG	Trial 1	Trial 2	Trial 3	AVG
0	275	143	170	196	1798	1715	2000	1837.66667	148	87	59	98
1	331	173	217	240.333	1721	1796	2000	1839	216	170	95	160.333
2	372	196	247	271.667	1722	1808	2000	1843.33333	279	220	123	207.333
3	404	214	268	295.333	1809	1847	2000	1885.33333	324	266	147	245.667
4	431	229	283	314.333	1840	1871	2000	1903.66667	404	316	174	298
5	454	242	295	330.333	1860	1937	2000	1932.33333	464	371	198	344.333
6	473	252	306	343.667	1885	1951	2000	1945.33333	533	429	225	395.667
7	491	262	316	356.333	1894	1990	2000	1961.33333	591	476	246	437.667
8	507	270	325	367.333	1972	1996	2000	1989.33333	646	541	276	487.667
9	521	278	334	377.667	1950	1999	2000	1983	706	593	304	534.333
10	536	284	341	387	1966	2000	2000	1988.66667	782	658	332	590.667
11	549	291	350	396.667	1970	2000	2000	1990	845	724	361	643.333
12	560	297	358	405	1979	2000	2000	1993	914	790	390	698
13	573	303	366	414	1899	2000	2000	1966.33333	1045	849	418	770.667
14	579	309	374	420.667	1918	2000	2000	1972.66667	1109	928	448	828.333
15	597	314	381	430.667	1970	2000	2000	1990	1143	976	477	865.333

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>		
	<i>At room temperature 31°C</i>	<i>In Ice cold water at 9°C</i>	<i>In Hot water at 75°C</i>
<i>0</i>	196	1837.66667	98
<i>1</i>	240.333	1839	160.333
<i>2</i>	271.667	1843.33333	207.333
<i>3</i>	295.333	1885.33333	245.667
<i>4</i>	314.333	1903.66667	298
<i>5</i>	330.333	1932.33333	344.333
<i>6</i>	343.667	1945.33333	395.667
<i>7</i>	356.333	1961.33333	437.667
<i>8</i>	367.333	1989.33333	487.667
<i>9</i>	377.667	1983	534.333
<i>10</i>	387	1988.66667	590.667
<i>11</i>	396.667	1990	643.333
<i>12</i>	405	1993	698
<i>13</i>	414	1966.33333	770.667
<i>14</i>	420.667	1972.66667	828.333
<i>15</i>	430.667	1990	865.333

Table 4 : Effect of temperature on light intensity of LIME-GREEN coloured glow stick

Time (in minutes)	Photo-resistance (k Ω)											
	At room temperature 31°C				In Ice cold water at 9°C				In Hot water at 75°C			
	Trial 1	Trial 2	Trial 3	AVG	Trial 1	Trial 2	Trial 3	AVG	Trial 1	Trial 2	Trial 3	AVG
0	61	42	50	51	687	493	484	554.666667	39	24	18	27
1	86	60	61	69	687	502	509	566	76	42	31	49.6667
2	101	71	72	81.3333	708	506	538	584	90	52	37	59.6667
3	119	84	81	94.6667	725	510	555	596.666667	118	62	48	76
4	128	88	86	100.667	732	516	561	603	127	73	56	85.3333
5	132	91	91	104.667	741	518	568	609	160	84	64	102.667
6	135	94	94	107.667	746	521	575	614	178	93	73	114.667
7	143	99	102	114.667	759	528	586	624.333333	199	104	82	128.333
8	145	101	106	117.333	765	532	593	630	225	118	91	144.667
9	148	103	154	135	771	536	600	635.666667	246	132	99	159
10	151	107	161	139.667	779	542	607	642.666667	268	146	108	174
11	157	111	168	145.333	790		547	614	650.333333	295	160	119
12	160	117	121	132.667	801	555	625	660.333333	312	172	128	204
13	170	117	125	137.333	814	562	634	670	340	185	138	221
14	174	120	128	140.667	824	569	641	678	373	202	149	241.333
15	177	122	130	143	829	574	647	683.333333	402	215	160	259

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>		
	<i>At room temperature 31°C</i>	<i>In Ice cold water at 9°C</i>	<i>In Hot water at 75°C</i>
<i>0</i>	51	554.666667	27
<i>1</i>	69	566	49.6667
<i>2</i>	81.3333	584	59.6667
<i>3</i>	94.6667	596.666667	76
<i>4</i>	100.667	603	85.3333
<i>5</i>	104.667	609	102.667
<i>6</i>	107.667	614	114.667
<i>7</i>	114.667	624.333333	128.333
<i>8</i>	117.333	630	144.667
<i>9</i>	135	635.666667	159
<i>10</i>	139.667	642.666667	174
<i>11</i>	145.333	650.333333	191.333
<i>12</i>	132.667	660.333333	204
<i>13</i>	137.333	670	221
<i>14</i>	140.667	678	241.333
<i>15</i>	143	683.333333	259

Table 5 : Effect of temperature on light intensity of YELLOW coloured glow stick

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>											
	<i>At room temperature 31°C</i>				<i>In Ice cold water at 9°C</i>				<i>In Hot water at 75°C</i>			
	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>
0	57	62	45	54.6667	528	540	353	473.667	20	16	16	17.3333
1	76	79	64	73	551	549	363	487.667	43	33	35	37
2	88	89	76	84.3333	573	549	367	496.333	61	45	47	51
3	98	97	85	93.3333	591	554	373	506	81	56	61	66
4	105	102	92	99.6667	610	562	379	517	95	65	75	78.3333
5	107	108	97	104	628	570	386	528	110	76	90	92
6	117	112	102	110.333	643	577	392	537.333	145	88	103	112
7	122	116	107	115	655	584	397	545.333	152	103	117	124
8	126	120	111	119	665	591	401	552.333	204	116	131	150.333
9	130	123	114	122.333	675	594	406	558.333	262	148	148	186
10	134	127	118	126.333	682	602	410	564.667	262	162	167	197
11	137	130	121	129.333	689	591	414	564.667	263	175	183	207
12	140	134	125	133	696	594	417	569	264	180	184	209.333
13	144	137	128	136.333	703	597	420	573.333	281	187	199	222.333
14	147	140	131	139.333	707	601	422	576.667	290	200	219	236.333
15	150	143	134	142.333	714	605	424	581	296	215	246	252.333

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>		
	<i>At room temperature 31°C</i>	<i>In Ice cold water at 9°C</i>	<i>In Hot water at 75°C</i>
0	54.6667	473.667	17.3333
1	73	487.667	37
2	84.3333	496.333	51
3	93.3333	506	66
4	99.6667	517	78.3333
5	104	528	92
6	110.333	537.333	112
7	115	545.333	124
8	119	552.333	150.333
9	122.333	558.333	186
10	126.333	564.667	197
11	129.333	564.667	207
12	133	569	209.333
13	136.333	573.333	222.333
14	139.333	576.667	236.333
15	142.333	581	252.333

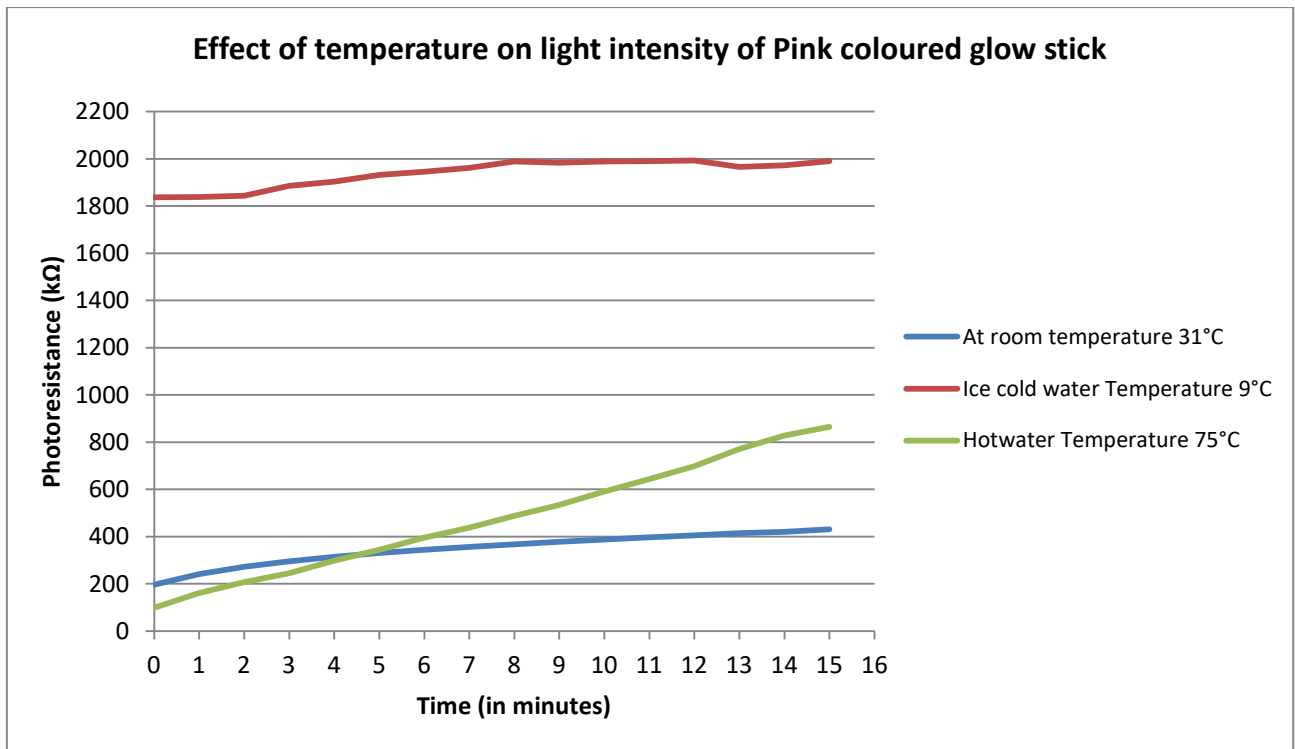
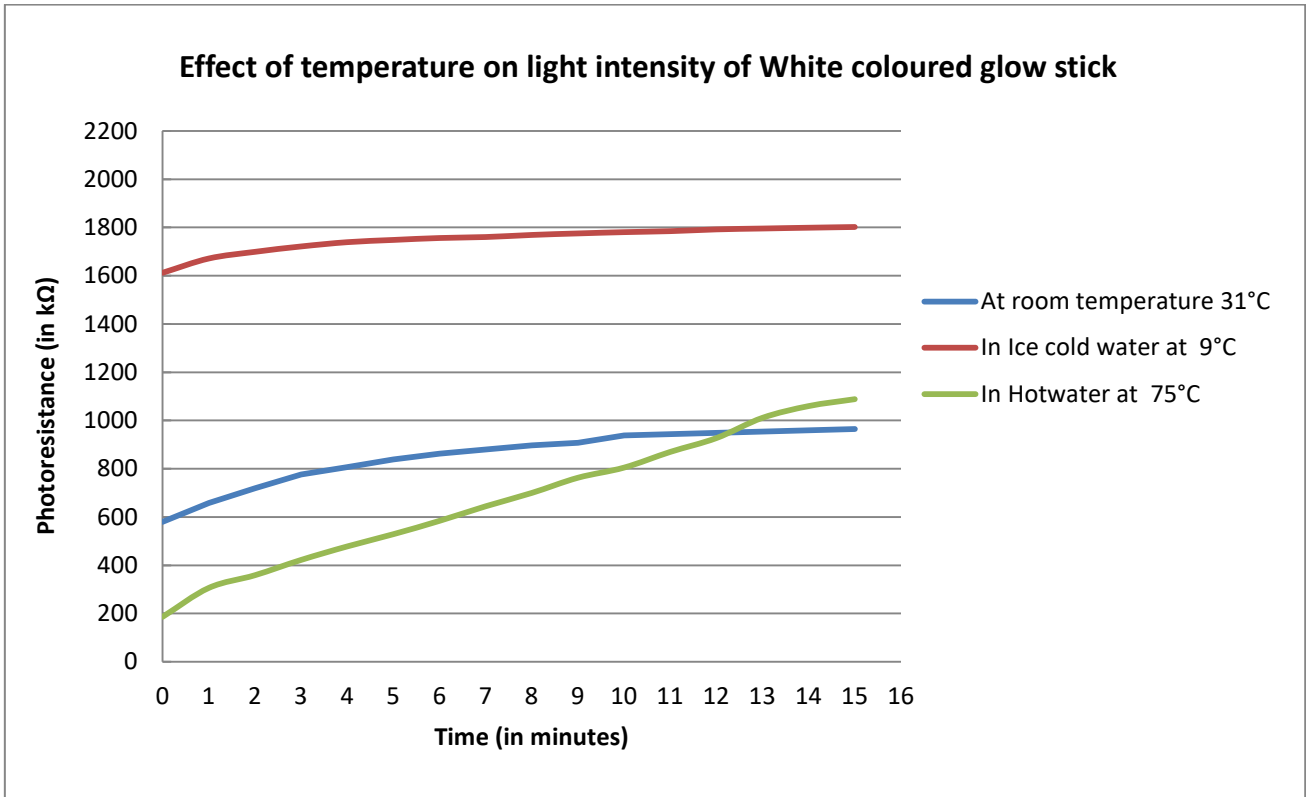
Table 6 : Effect of temperature on light intensity of ORANGE coloured glow stick

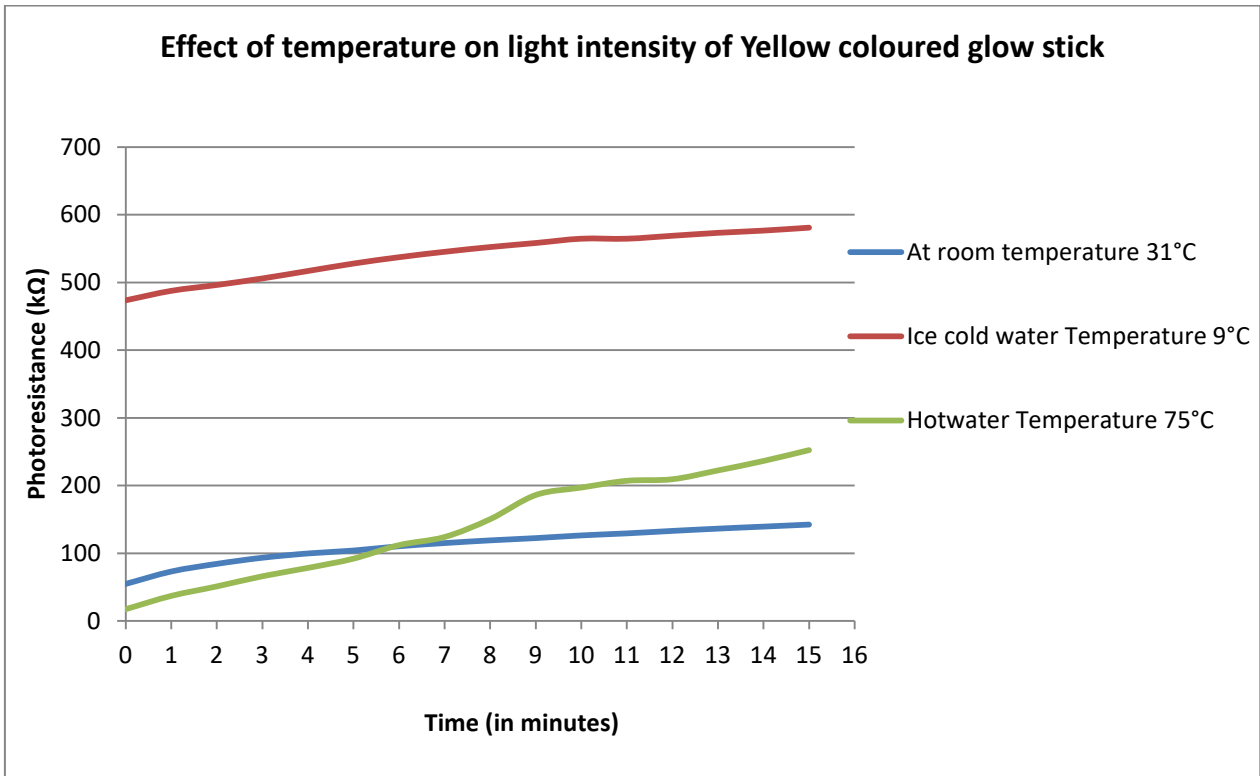
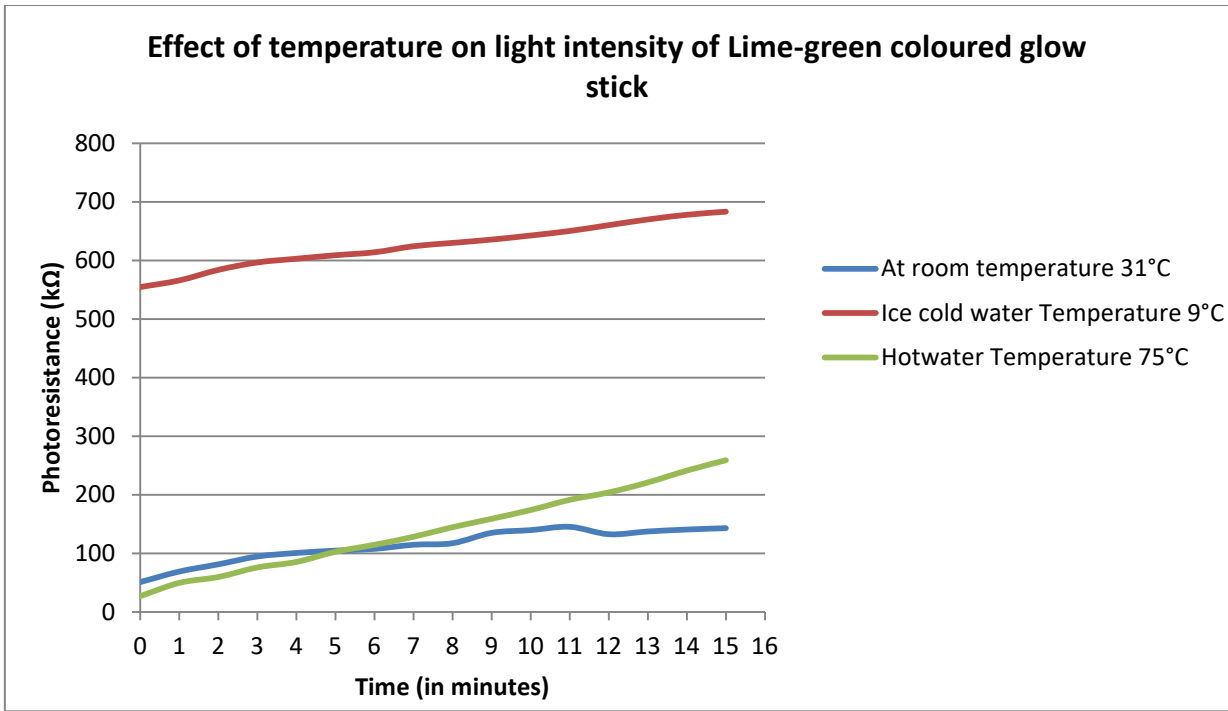
<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>											
	<i>At room temperature 31°C</i>				<i>In Ice cold water at 9°C</i>				<i>In Hot water at 75°C</i>			
	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 3</i>	<i>AVG</i>
0	94	114	56	88	502	1101	569	724	19	24	19	20.6667
1	106	137	80	107.667	547	1189	599	768.333	65	61	38	54.6667
2	114	152	93	119.667	552	1210	616	787	89	78	49	72
3	122	165	102	129.667	555	1240	635	810	121	94	60	91.6667
4	127	176	109	137.333	557	1270	665	830.667	140	125	75	113.333
5	132	185	115	144	565	1302	694	853.667	162	153	87	134
6	137	193	119	149.667	581	1329	720	876.667	170	179	127	158.667
7	141	200	123	154.667	591	1351	743	895	187	208	161	185.333
8	144	206	127	159	596	1369	761	908.667	208	231	193	210.667
9	148	213	131	164	597	1374	773	914.667	293	259	213	255
10	151	218	134	167.667	601	1387	786	924.667	325	290	226	280.333
11	155	224	137	172	605	1395	795	931.667	358	321	268	315.667
12	159	230	140	176.333	609	1403	804	938.667	385	349	296	343.333
13	162	236	143	180.333	611	1406	812	943	425	387	348	386.667
14	164	240	146	183.333	615	1410	819	948	425	404	351	393.333
15	168	246	149	187.667	617	1414	825	952	456	440	394	430

<i>Time (in minutes)</i>	<i>Photo-resistance (kΩ)</i>		
	<i>At room temperature 31°C</i>	<i>In Ice cold water at 9°C</i>	<i>In Hot water at 75°C</i>
<i>0</i>	88	724	20.6667
<i>1</i>	107.667	768.333	54.6667
<i>2</i>	119.667	787	72
<i>3</i>	129.667	810	91.6667
<i>4</i>	137.333	830.667	113.333
<i>5</i>	144	853.667	134
<i>6</i>	149.667	876.667	158.667
<i>7</i>	154.667	895	185.333
<i>8</i>	159	908.667	210.667
<i>9</i>	164	914.667	255
<i>10</i>	167.667	924.667	280.333
<i>11</i>	172	931.667	315.667
<i>12</i>	176.333	938.667	343.333
<i>13</i>	180.333	943	386.667
<i>14</i>	183.333	948	393.333
<i>15</i>	187.667	952	430

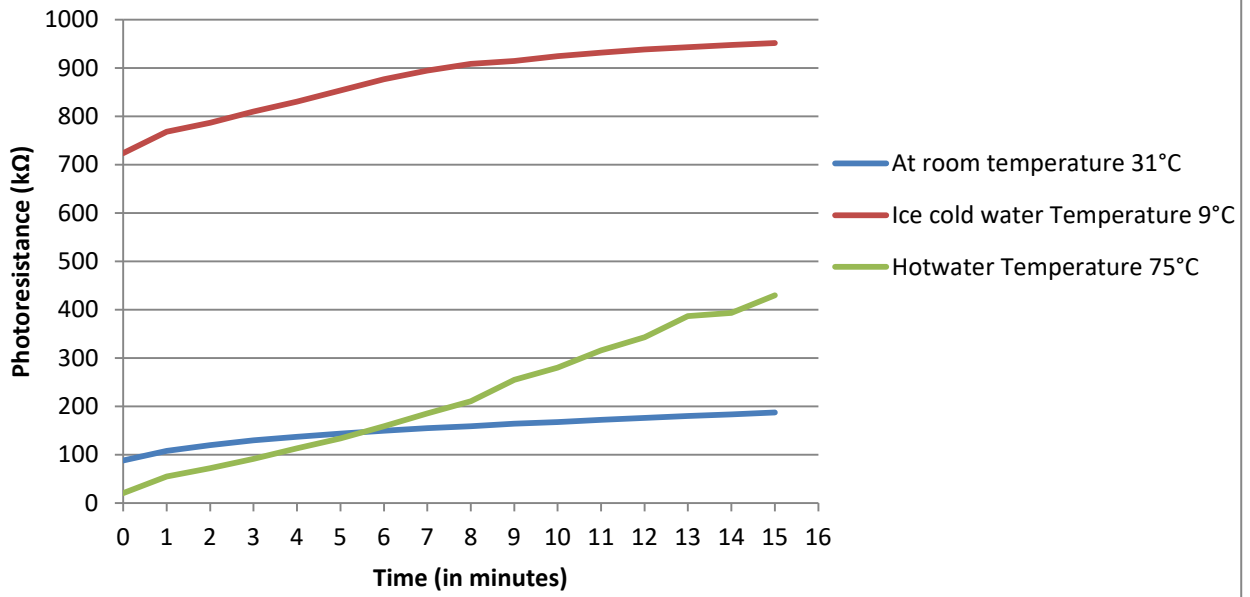
Graphical Representation

THE EFFECT OF TEMPERATURE ON LIGHT INTENSITY OF DIFFERENT COLOURED GLOW STICKS

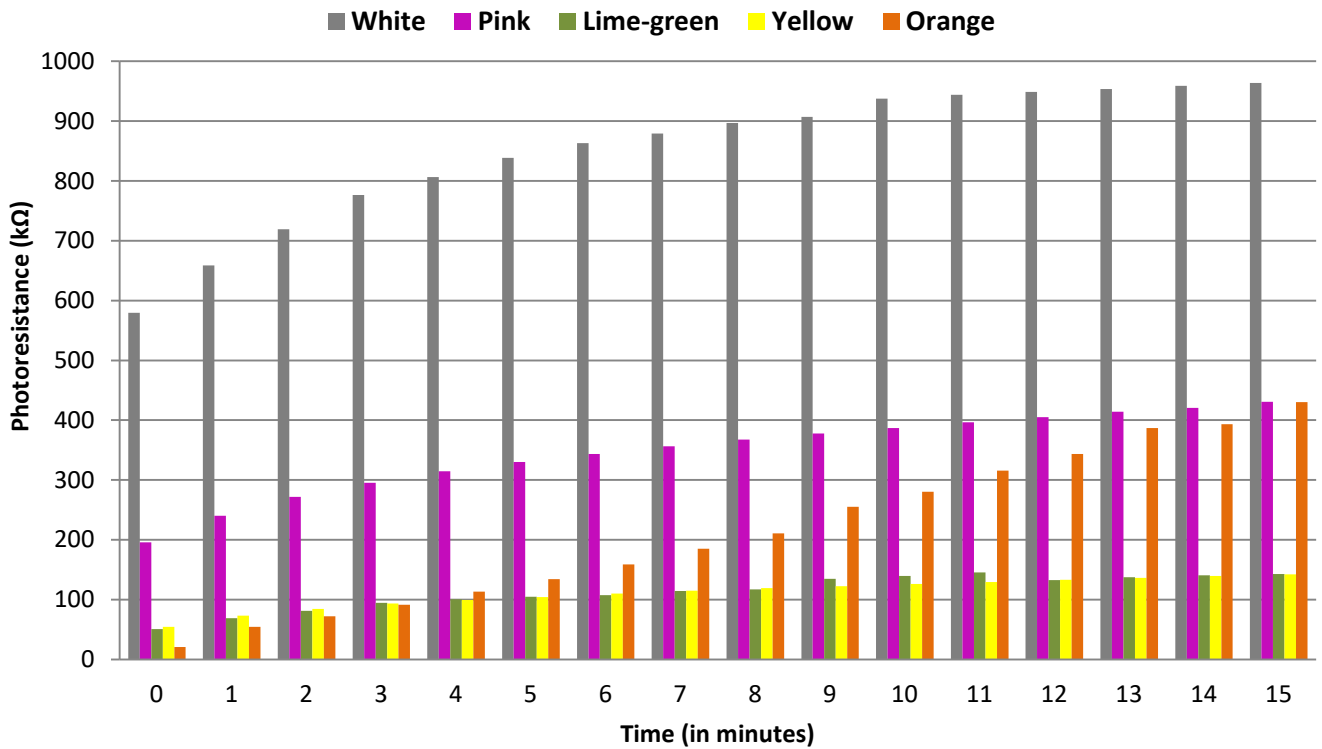




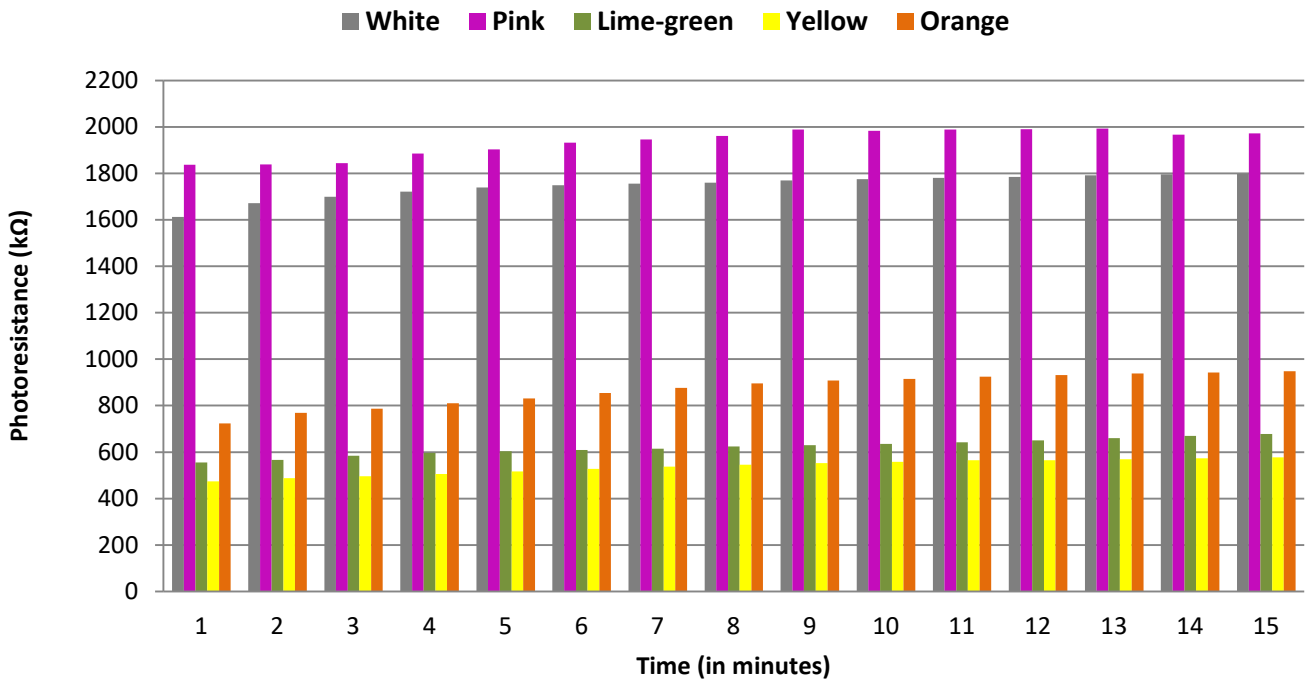
Effect of temperature on light intensity of Orange coloured glow stick



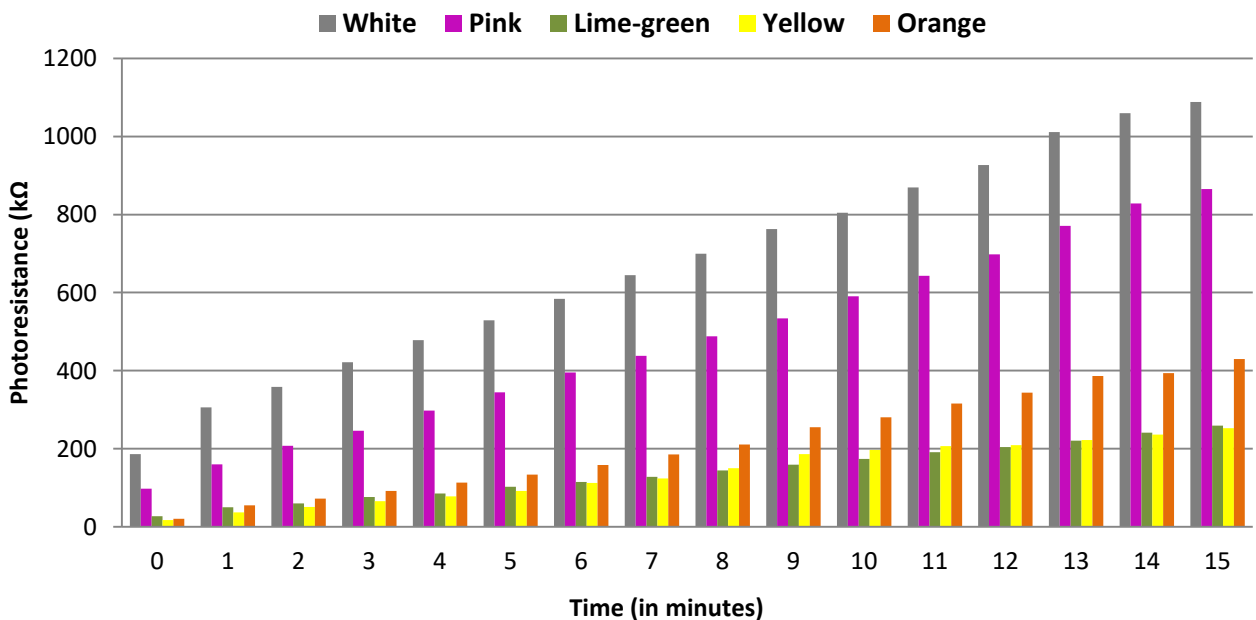
THE EFFECT OF ROOM TEMPERATURE ON LIGHT INTENSITY OF DIFFERENT COLOURED GLOW STICKS



THE EFFECT OF ICE COLD TEMPERATURE ON LIGHT INTENSITY OF DIFFERENT COLOURED GLOW STICKS



THE EFFECT OF HOT TEMPERATURE ON LIGHT INTENSITY OF DIFFERENT COLOURED GLOW STICKS



RESULTS AND DISCUSSION

- In the baseline test, the resistance of photo-resistor with the closed lid was above 2000kΩ. I used this as confirmation test value for knowing the baseline of the photo-resistor I had taken.

A. Time Vs Light intensity

- By plotting the data, I got a *more or less a straight line, away from x-axis* which indicates the photo-resistance is directly proportional to the time.
- Increasing photo-resistance implies decreasing light intensity. Hence, *light intensity decays over time.*

B. Colour Vs Light intensity

- Comparing all, *Yellow has more light intensity in all the temperature.*

- *Increasing order of light intensity at room temperature 21°C:*

White < Pink < Orange < Lime-green < Yellow

- *Increasing order of light intensity in ice cold water at 9°C :*

Pink < White < Orange < Lime-green < Yellow

- *Increasing order of light intensity in hot water at 75°C:*

White < Pink < Orange < Lime-green < Yellow

C. Temperature Vs Light intensity

- I observed the greater increase in the photo-resistance in ice cold water from that of hot and room temperatures, for all the colours both at initial and final time point.
- Comparing hot and room temperature, resistance was initially low at hot which means high light intensity but gradually increased after 15minutes, whereas at room temperature, resistance maintains an approximate value. It may be due to the loss of heat. From this, I formulated the fact that *increasing temperature actually increases light intensity* unless and until the heat is maintained.
- The range of multimeter I had was 2000kΩ. For the, I have considered display readings 1 as 2000kΩ and plotted my data.
- Some glow sticks may have manufacturing defect, so I have taken three trials for promising results.

CONCLUSION

My hypothesis, “Cold temperature increases the Light intensity of glow sticks” has not been proved. I found that increasing temperature increases the light intensity.

FUTURE ENHANCEMENT

- I want to continue my experiment with the apt range of multimeter ($>2000k\Omega$).
- I should update my setup so that the temperature drop/rise can also be noted along with the resistance so as to determine the particular point of temperature at which the resistances increase/decrease.
- Furthermore, I want to compare among different hot temperatures ranging from 50°C to 100°C for colours other than the ones I have used.

APPLICATION

- The science behind chemiluminescence could be possibly utilized for future light generating mechanisms.
- Chemiluminescence is a non-radioactive, chemical reaction which produces a white light. It finds its application in *chemiluminescent western blotting*, where chemiluminescence occurs when a chemiluminescent substrate attached to an antibody bound to the target protein undergoes an enzyme-based reaction.
- Chemiluminescence is more sensitive than colorimetric protein detection because the antibody targets the protein specifically and the enzyme amplifies the light signal.

Since chemiluminescent molecules produce so many photons of light, with the right types of image capture technology, *chemiluminescence can be used to detect proteins that are expressed at low levels*. Since chemiluminescent light is generated internally, protein sample won't be damaged, and can be re-used.

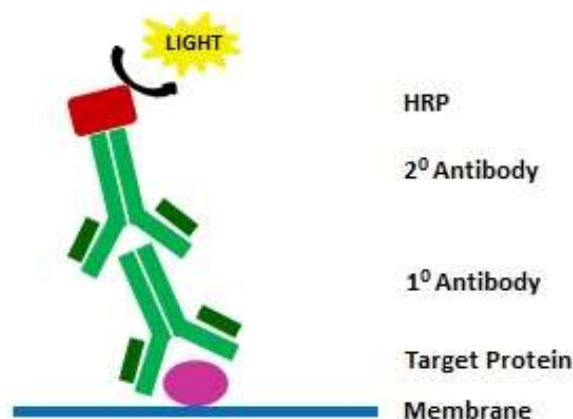


Figure 1- Chemiluminescence western blot
Primary antibody (1°) binds to a target protein on the membrane. The membrane is then incubated with a chemiluminescence substrate e.g. HRP. An enzymatic reaction occurs producing light which is detected by a CCD camera.

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